

# Two Switchable Star-Shaped [1](*n*)Rotaxanes with Different Multibranching Cores

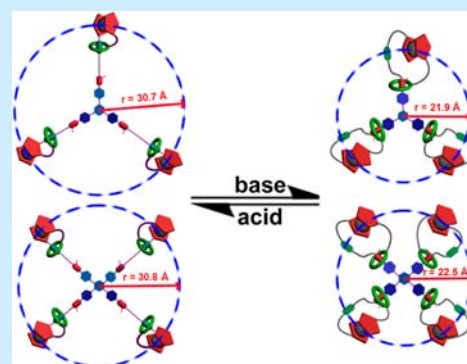
Hong Li,<sup>†</sup> Xin Li,<sup>‡</sup> Hans Ågren,<sup>‡</sup> and Da-Hui Qu<sup>\*,†</sup>

<sup>†</sup>Key Laboratory for Advanced Materials and Institute of Fine Chemicals, East China University of Science & Technology, Shanghai 200237, P. R. China

<sup>‡</sup>Division of Theoretical Chemistry and Biology, School of Biotechnology, KTH Royal Institute of Technology, SE-10691 Stockholm, Sweden

## Supporting Information

**ABSTRACT:** Two novel star-shaped [1](*n*)rotaxanes with three and four identical [1]rotaxane arms but different multibranching cores were designed, synthesized, and well-characterized. In the two systems, external base–acid stimuli result in the uniform relative mechanical movement of the macrocyclic rings and threads of their [1]rotaxane arms. The energy-minimized structures of the two rotaxanes in different states were obtained using molecular dynamics simulations in acetone solution, suggesting the construction of more sophisticated molecular machines mimicking the extension and contraction motions.



Along with the rapid development of supramolecular chemistry, much attention has been paid to the design and construction of increasingly sophisticated mechanically interlocked molecules (MIMs)<sup>1</sup> possessing special structures and shapes. Until now, many of these MIMs and other supramolecular assemblies with special shapes, such as [2](2)rotaxane,<sup>2</sup> handcuff catenane,<sup>3</sup> molecular elevators,<sup>4</sup> molecular necklace,<sup>5</sup> molecular walkers,<sup>6</sup> and molecular muscles,<sup>7</sup> have been designed and constructed as potential components of molecular machinery.<sup>8</sup> Currently, there is a growing interest in [1]rotaxanes, as new types of MIMs, because of their intriguing topological structure<sup>9</sup> and their application as molecular muscle.<sup>10</sup> However, the construction of highly ordered and/or symmetrical assemblies containing multiple mechanically interlocked rotaxane building blocks still remains a challenge due to the increased complexity of their chemical structures.<sup>11</sup> This inspired us to construct multibranching mechanically interlocked molecular systems employing [1]rotaxane building blocks as arms.

In this paper, two star-shaped [1](*n*)rotaxanes (here [1] indicates the number of molecular components and (*n*) stands for the number of interlocked structures involved<sup>2b,c</sup>), a novel tri- and tetrabranching [1]rotaxane, having identical [1]rotaxane arms but different cores, were designed, synthesized, and characterized. As shown in Figure 1, the [1]rotaxane arm bears a ferrocene (Fc)<sup>12</sup> unit that connects a dibenzo-24-crown-8 (DB24C8) macrocycle and a rodlike part through its cyclopentadienyl (Cp) rings. The rodlike part has two distinguishable recognition stations for DB24C8, namely a dibenzylammonium (DBA)<sup>13</sup> station and a *N*-methyltriaxolium

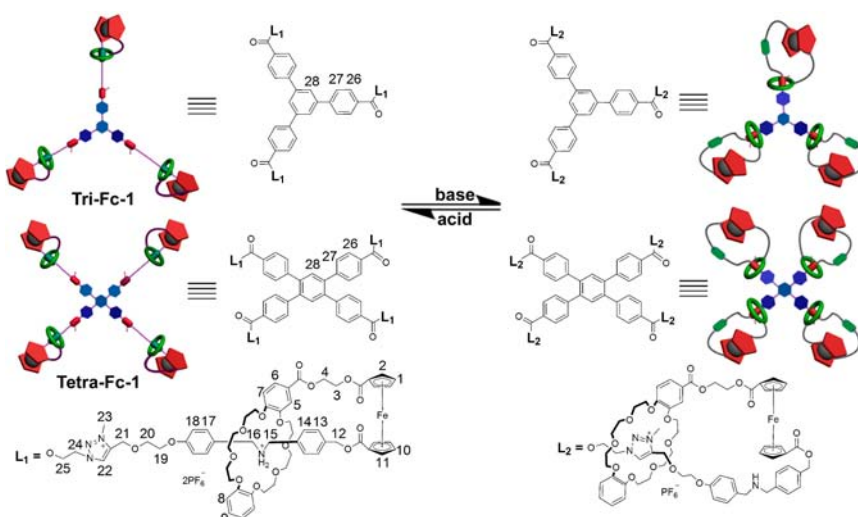
(MTA)<sup>14</sup> station. Because of their rigid structures and symmetrical shapes,<sup>15</sup> a 1,3,5-triphenylene and a 1,2,4,5-tetraphenylene core were used for the tribranching [1](3)rotaxane **Tri-Fc-1** and for the tetrabranching [1](4)rotaxane **Tetra-Fc-1**, respectively.

In these two [1](*n*)rotaxane systems, the uniform relative mechanical movements of their macrocyclic rings and threads can be driven by external base–acid stimuli, confirmed by <sup>1</sup>H NMR spectroscopy. Furthermore, the energy-minimized structures of the two [1](*n*)rotaxanes in different states have been investigated by molecular dynamics (MD) simulations<sup>16</sup> in acetone solution to gain further insight into the behavior of these highly complex and symmetrical [1]rotaxane systems. The calculated percentage changes for **Tri-Fc-1** and **Tetra-Fc-1** in their two states are about 28.7% and 26.9%, respectively, which are similar to the percentage change (~27%) in human muscle. This indicates that these systems can mimic the extension and contraction motions of skeletal muscle while undergoing stimuli-induced structural changes.

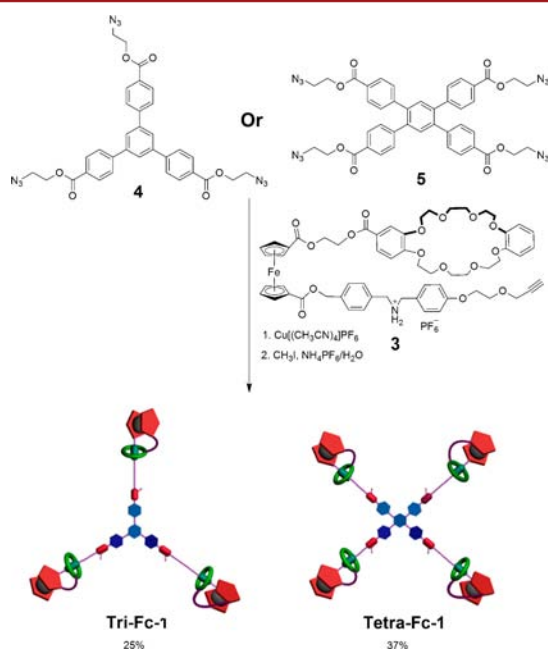
The syntheses of the target tribranching [1](3)rotaxane **Tri-Fc-1** and tetrabranching [1](4)rotaxane **Tetra-Fc-1** are shown in Figure 2. First, we designed and synthesized compounds **4** and **5** (Figure 2) the tri- and tetrabranching cores that are key intermediates toward the target rotaxanes (their detailed syntheses are shown in Scheme S1, Supporting Information). In the next step, we synthesized compound **3**<sup>17</sup> comprising two subunits, a DB24C8 macrocycle and a rodlike part with a DBA

Received: August 20, 2014

Published: September 10, 2014



**Figure 1.** Switching processes and the schematic representations of the novel tri- and tetrabranch [1](*n*)rotaxanes **Tri-Fc-1** and **Tetra-Fc-1**, respectively.



**Figure 2.** Syntheses of the target [1](*n*)rotaxanes **Tri-Fc-1** and **Tetra-Fc-1**.

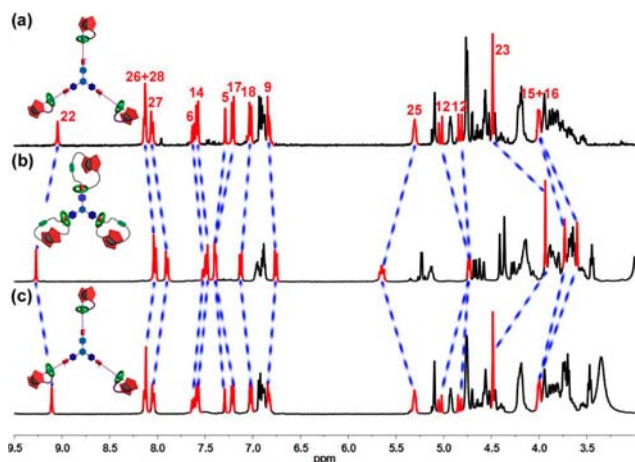
station and a terminal alkyne, both attached to the Cp rings of a central ferrocene unit via ester bonds. Finally, as shown in Figure 2, the self-complexing [1]pseudorotaxane **3** in  $\text{CH}_2\text{Cl}_2$  was coupled with the azides of the tri- and tetrabranch compounds **4** and **5** using  $\text{Cu}(\text{CH}_3\text{CN})_4\text{PF}_6$  as catalyst through a “threading-followed-by-stoppering” approach<sup>18</sup> to obtain intermediates **Tri-Fc-2** and **Tetra-Fc-2** (Figure S2, Supporting Information). Subsequently, **Tri-Fc-2** and **Tetra-Fc-2** were treated with  $\text{CH}_3\text{I}$  followed by anion exchange in acetone solution with saturated  $\text{NH}_4\text{PF}_6$  to get the target rotaxanes **Tri-Fc-1** and **Tetra-Fc-1** in 25% and 37% yield, respectively (Figure 2).

Characterization of the target tribranched [1](3)rotaxane **Tri-Fc-1** and tetrabranch [1](4)rotaxane **Tetra-Fc-1** was performed by  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectroscopies and HR-ESI mass spectrometry. The HR-ESI mass spectrum of the

tribranched [1](3)rotaxane **Tri-Fc-1** displays the strongest peak at  $m/z$  1070.8320, corresponding to a species having lost four  $\text{PF}_6^-$  counterions, i.e.,  $[\text{M} - 4\text{PF}_6]^{4+}$ , in agreement with the calculated value of 1070.8333 for  $[\text{C}_{213}\text{H}_{234}\text{N}_{12}\text{O}_{54}\text{Fe}_3\text{P}_2\text{F}_{12}]$ . Two other strong peaks at  $m/z$  1476.0929 and 827.6719, corresponding to species having lost three and five  $\text{PF}_6^-$  counterions ( $[\text{M} - 3\text{PF}_6]^{3+}$  and  $[\text{M} - 5\text{PF}_6]^{5+}$ ), respectively, are also observed. Furthermore, the HR-ESI mass spectrum of tetrabranch [1](4)rotaxane **Tetra-Fc-1** also shows three intense peaks at  $m/z$  1469.7020, 1146.7595, and 931.4730, corresponding to species having lost four, five, and six  $\text{PF}_6^-$  counterions, that is  $[\text{M} - 4\text{PF}_6]^{4+}$ ,  $[\text{M} - 5\text{PF}_6]^{5+}$ , and  $[\text{M} - 6\text{PF}_6]^{6+}$ , respectively.

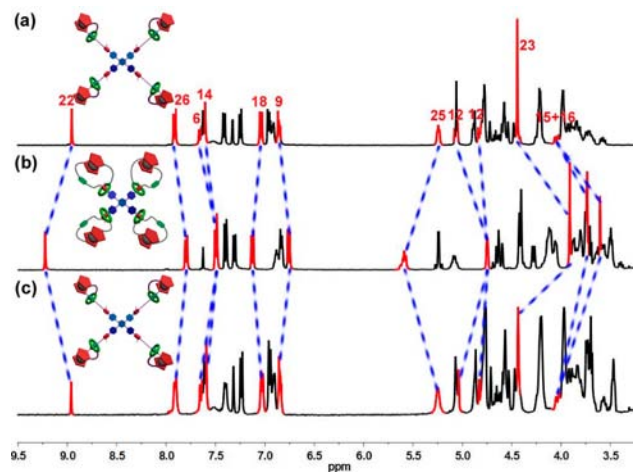
Next, we investigated the uniform relative mechanical movements of the macrocyclic DB24C8 rings and threads in response to external base–acid stimuli using  $^1\text{H}$  NMR spectroscopy. In the tribranched [1](3)rotaxane **Tri-Fc-1** system, the MTA recognition stations were migrated into the DB24C8 rings upon addition of 4.0 equiv of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) to its  $\text{CD}_3\text{COCD}_3$  solution because of the deprotonation of the DBA stations by the excess DBU. These structural changes were accompanied by some obvious  $^1\text{H}$  NMR spectral changes (Figure 3). As shown in Figure 3b, the protons  $\text{H}_{15}$ ,  $\text{H}_{16}$  on the DBA station were split and shifted upfield with  $\Delta\delta$  of  $-0.28$ ,  $-0.41$  ppm, respectively. Similarly, the phenyl protons  $\text{H}_5$ ,  $\text{H}_6$ ,  $\text{H}_9$  on the macrocycle DB24C8 were also changed (0.10,  $-0.13$ , and  $-0.08$  ppm, respectively). Meanwhile, the protons  $\text{H}_{22}$  and  $\text{H}_{23}$  on the MTA station were shifted dramatically with  $\Delta\delta$  of 0.22 and  $-0.56$  ppm, respectively. It should be noted that the protons  $\text{H}_{12}$  near the ferrocene were merged into a single doublet from the original two doublets ( $-0.10$  and  $-0.31$  ppm, respectively). All this evidence indicated that the three MTA stations all migrated into the DB24C8 rings upon addition of excess DBU.

The generated dibenzylamine center could be reprotonated again by the addition of 8.0 equiv of  $\text{CF}_3\text{COOH}$  (TFA), which led to the regeneration of the original  $^1\text{H}$  NMR spectrum (Figure 3c). Collectively, these data demonstrate that in the tribranched [1](3)rotaxane **Tri-Fc-1** system the uniform relative mechanical movement of the DB24C8 rings and threads can be reversibly switched in response to external



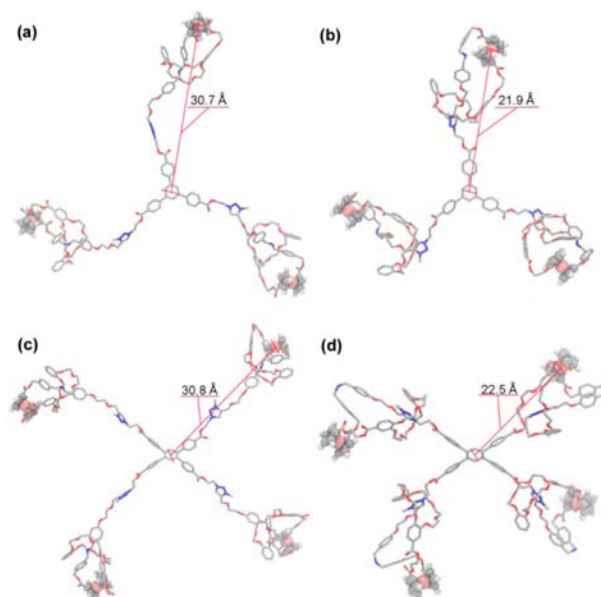
**Figure 3.** Partial  $^1\text{H}$  NMR spectra (400 MHz,  $\text{CD}_3\text{COCD}_3$ , 298 K,  $2.0 \times 10^{-3}$  M) of (a)  $[1](3)$ rotaxane **Tri-Fc-1**, (b) deprotonation upon addition of 4.0 equiv of DBU to sample a, and (c) reprotation upon addition of 8.0 equiv of TFA to sample b.

base–acid stimuli. Similarly, in the tetrabranch  $[1](4)$ -rotaxane **Tetra-Fc-1** system, the uniform and reversible relative mechanical movement of its macrocyclic rings and threads in response to external base–acid stimuli has the same trend as that observed in the tribranched  $[1](3)$ rotaxane **Tri-Fc-1** system (Figure 4).



**Figure 4.** Partial  $^1\text{H}$  NMR spectra (400 MHz,  $\text{CD}_3\text{COCD}_3$ , 298 K,  $2.0 \times 10^{-3}$  M) of (a)  $[1](4)$ rotaxane **Tetra-Fc-1**, (b) deprotonation upon addition of 5.0 equiv of DBU to sample a, and (c) reprotation upon addition of 10.0 equiv of TFA to sample b.

To further recognize the structural differences of the two  $[1](n)$ rotaxanes **Tri-Fc-1** and **Tetra-Fc-1** in two states, their energy-minimized structures in different states have been investigated by MD simulations in acetone solution. It can be seen from Figure 5 that the two  $[1](n)$ rotaxanes in original states have their three and/or four crown ether macrocycles encircled at the protonated amine groups and feature extended conformations, considering the flexibility of the rotaxane arms; this illustrates the near-maximal separation between the central aromatic core and the peripheral ferrocene units. The average distances of the extended conformations in  $[1](n)$ rotaxanes **Tri-Fc-1** and **Tetra-Fc-1** between the central phenyl ring and the Fe(II) atom are about 30.7 and 30.8 Å, respectively (Figure



**Figure 5.** Snapshots for the structures of the  $[1](n)$ rotaxanes **Tri-Fc-1** and **Tetra-Fc-1** at their two states, extracted from molecular dynamics trajectories. The optimized conformations at different states were obtained through molecular dynamics simulations employing the general Amber force field.

5a,c). However, upon addition of excess DBU, all of their three and/or four crown ether macrocycles encircle on the MTA stations, and the optimized structures exhibit contraction motions and partial folding of the rotaxane arms, leading to a shortened distance between the central phenyl ring and the Fe(II) atoms of 21.9 and 22.5 Å, respectively (Figure 5b,d). As such, a fair estimation of the percentage changes of dimension in each arm of  $[1](n)$ rotaxanes **Tri-Fc-1** and **Tetra-Fc-1** in two different states is obtained as 28.7% for the former and 26.9% for the latter, which are similar to the percentage change ( $\sim 27\%$ ) in human muscle. These results confirmed that the two  $[1](n)$ rotaxanes systems could mimic the extension and contraction motions in response to external base–acid stimuli.

In summary, we have designed and constructed two switchable star-shaped  $[1](n)$ rotaxanes by introducing a tribranched and a tetrabranch core that connects  $[1]$ rotaxane building blocks. The external base–acid stimuli resulted in the reversible relative mechanical movement between the macrocycle and the thread components in each arm of the two star-shaped systems, ultimately generating uniform extension and contraction of the whole molecular system. The energy minimized structures obtained from molecular dynamics simulations showed that the percentage change of the dimension in each arm is very similar to that in human muscle. These kinds of systems can pave the way for designing more complicated molecular and supramolecular assemblies with increased structural complexity and specific functions.

## ■ ASSOCIATED CONTENT

### Supporting Information

Full experimental procedures and characterization data ( $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR, and HR-ESI) of all new compounds including tribranched  $[1](3)$ rotaxane **Tri-Fc-1**, tetrabranch  $[1](4)$ rotaxane **Tetra-Fc-1**, and key intermediates. This

material is available free of charge via the Internet at <http://pubs.acs.org>.

## AUTHOR INFORMATION

### Corresponding Author

\*E-mail: [dahui\\_qu@ecust.edu.cn](mailto:dahui_qu@ecust.edu.cn).

### Notes

The authors declare no competing financial interest.

## ACKNOWLEDGMENTS

We thank the NSFC/China (21272073, 21190033) and National Basic Research 973 Program (2011CB808400), the Fok Ying Tong Education Foundation (121069), the Fundamental Research Funds for the Central Universities, the Innovation Program of Shanghai Municipal Education Commission, and the Science Fund for Creative Research Groups (No. 21421004) for financial support. We thank Dr. Gábor London from the University of Szeged for helpful discussions.

## REFERENCES

- (1) (a) Kinbara, K.; Aida, T. *Chem. Rev.* **2005**, *105*, 1377–1400. (b) Tian, H.; Wang, Q.-C. *Chem. Soc. Rev.* **2006**, *35*, 361–374. (c) Saha, S.; Stoddart, J. F. *Chem. Soc. Rev.* **2007**, *36*, 77–92. (d) Balzani, V.; Credi, A.; Venturi, M. *Molecular Devices and Machines - Concepts and Perspectives for the Nanoworld*; Wiley-VCH: Weinheim, 2008. (e) Balzani, V.; Credi, A.; Venturi, M. *Chem. Soc. Rev.* **2009**, *38*, 1542–1550. (f) Qu, D.-H.; Tian, H. *Chem. Sci.* **2011**, *2*, 1011–1015. (g) Evans, N. H.; Beer, P. D. *Chem. Soc. Rev.* **2014**, *43*, 4658–4683. (h) Ceroni, P.; Credi, A.; Venturi, M. *Chem. Soc. Rev.* **2014**, *43*, 4068–4083.
- (2) (a) Ciao, R.; Talotta, C.; Gaeta, C.; Margarucci, L.; Casapullo, A.; Neri, P. *Org. Lett.* **2013**, *15*, 5694–5697. (b) Ma, Y.-X.; Meng, Z.; Chen, C.-F. *Org. Lett.* **2014**, *16*, 1860–1863. (c) Li, H.; Li, X.; Wu, Y.; Ågren, H.; Qu, D.-H. *J. Org. Chem.* **2014**, *79*, 6996–7004.
- (3) Hartlie, K. J.; Blackburn, A. K.; Schneebeli, S. T.; Forgan, R. S.; Sarjeant, A. A.; Stern, C. L.; Cao, D.; Stoddart, J. F. *Chem. Sci.* **2014**, *5*, 90–100.
- (4) (a) Badjic, J. D.; Balzani, V.; Credi, A.; Silvi, S.; Stoddart, J. F. *Science* **2004**, *303*, 1845–1849. (b) Badjic, J. D.; Ronconi, C. M.; Stoddart, J. F.; Balzani, V.; Silvi, S.; Credi, A. *J. Am. Chem. Soc.* **2006**, *128*, 1489–1499.
- (5) (a) Roh, S.-G.; Park, K.-M.; Park, G.-J.; Sakamoto, S.; Yamaguchi, K.; Kim, K. *Angew. Chem., Int. Ed.* **1999**, *38*, 637–641. (b) Kim, K. *Chem. Soc. Rev.* **2002**, *31*, 96–107. (c) Chang, C.-F.; Chuang, C.-J.; Lai, C.-C.; Liu, Y.-H.; Peng, S.-M.; Chiu, S.-H. *Angew. Chem., Int. Ed.* **2012**, *51*, 10094–10098.
- (6) (a) Delius, M. V.; Geertsema, E. M.; Leigh, D. A. *Nat. Chem.* **2009**, *2*, 96–101. (b) Lewandowski, B.; Bo, G. D.; Ward, J. W.; Pappmeyer, M.; Kuschel, S.; Aldegunde, M. J.; Gramlich, P. M. E.; Heckmann, D.; Goldup, S. M.; D'Souza, D. M.; Fernandes, A. E.; Leigh, D. A. *Science* **2013**, *339*, 189–193. (c) Qu, D.-H.; Tian, H. *Chem. Sci.* **2013**, *4*, 3031–3035.
- (7) (a) Jiménez, M. C.; Dietrich-Buchecker, C.; Sauvage, J.-P. *Angew. Chem., Int. Ed.* **2000**, *39*, 3284–3287. (b) Chuang, C.-J.; Li, W.-S.; Lai, C.-C.; Liu, Y.-H.; Peng, S.-M.; Chao, I.; Chiu, S.-H. *Org. Lett.* **2009**, *11*, 385–388. (c) Ma, X.; Tian, H. *Chem. Soc. Rev.* **2010**, *39*, 70–80. (d) Du, G.-Y.; Moulin, E.; Jouault, N.; Buhler, E.; Giuseppone, N. *Angew. Chem., Int. Ed.* **2012**, *51*, 12504–12508. (e) Romuald, C.; Ardá, A.; Clavel, C.; Jiménez-Barbero, J.; Coutrot, F. *Chem. Sci.* **2012**, *3*, 1851–1857.
- (8) (a) Coskun, A.; Friedman, D. C.; Li, H.; Patel, K.; Khatib, H. A.; Stoddart, J. F. *J. Am. Chem. Soc.* **2009**, *131*, 2493–2495. (b) Mateo-Alonso, A. *Chem. Commun.* **2010**, *46*, 9089–9099. (c) Coskun, A.; Banaszak, M.; Astumian, R. D.; Stoddart, J. F.; Grzybowski, B. A. *Chem. Soc. Rev.* **2012**, *41*, 19–30. (d) Neal, E. A.; Goldup, S. M. *Chem. Commun.* **2014**, *50*, 5128–5142.
- (9) (a) Hiratani, K.; Kaneyama, M.; Nagawa, Y.; Koyama, E.; Kaneshato, M. *J. Am. Chem. Soc.* **2004**, *126*, 13568–13569. (b) Zhou, Q.-Z.; Wei, P.-F.; Zhang, Y.-Y.; Yu, Y.-H.; Yan, X.-Z. *Org. Lett.* **2013**, *15*, 5350–5353.
- (10) (a) Gao, C.; Ma, X.; Zhang, Q.; Wang, Q.-C.; Qu, D.-H.; Tian, H. *Org. Biomol. Chem.* **2011**, *9*, 1126–1132. (b) Li, H.; Zhang, J.-N.; Zhou, W.; Zhang, H.; Zhang, Q.; Qu, D.-H.; Tian, H. *Org. Lett.* **2013**, *15*, 3070–3073.
- (11) (a) Zhang, Z.-J.; Zhang, H.-Y.; Wang, H.; Liu, Y. *Angew. Chem., Int. Ed.* **2011**, *50*, 10834–10838. (b) Zhang, H.; Liu, Q.; Li, J.; Qu, D.-H. *Org. Lett.* **2013**, *15*, 338–341. (c) Li, Z.-Y.; Liu, G.-X.; Xue, W.; Wu, D.; Yang, Y.-W.; Wu, J.-S.; Liu, S.-H.; Yoon, J.; Yin, J. *J. Org. Chem.* **2013**, *78*, 11560–11570.
- (12) (a) Mateo-Alonso, A.; Ehli, C.; Rahman, G. M. A.; Guldi, D. M.; Fioravanti, G.; Marcaccio, M.; Paolucci, F.; Prato, M. *Angew. Chem., Int. Ed.* **2007**, *46*, 3521–3525. (b) Mateo-Alonso, A.; Ehli, C.; Guldi, D. M.; Prato, M. *J. Am. Chem. Soc.* **2008**, *130*, 14938–14939. (c) Zhang, D.; Zhang, Q.; Sua, J.-H.; Tian, H. *Chem. Commun.* **2009**, 1700–1702. (d) Iordache, A.; Oltean, M.; Milet, A.; Thomas, F.; Baptiste, B.; Saint-Aman, E.; Bucher, C. *J. Am. Chem. Soc.* **2012**, *134*, 2653–2671.
- (13) (a) Jiang, Y.; Guo, J.-B.; Chen, C.-F. *Org. Lett.* **2010**, *12*, 4248–4251. (b) Clavel, C.; Romuald, C.; Brabet, E.; Coutrot, F. *Chem.—Eur. J.* **2013**, *19*, 2982–2989. (c) Zhou, W.; Guo, Y.-J.; Qu, D.-H. *J. Org. Chem.* **2013**, *78*, 590–596.
- (14) (a) Coutrot, F.; Romuald, C.; Busseron, E. *Org. Lett.* **2008**, *10*, 3741–3744. (b) Zhang, H.; Kou, X.-X.; Zhang, Q.; Qu, D.-H.; Tian, H. *Org. Biomol. Chem.* **2011**, *9*, 4051–4056. (c) Zhang, H.; Zhou, B.; Li, H.; Qu, D.-H.; Tian, H. *J. Org. Chem.* **2013**, *78*, 2091–2098. (d) Zhou, W.; Wu, Y.; Zhai, B.-Q.; Wang, Q.-C.; Qu, D.-H. *RSC Adv.* **2014**, *4*, 5148–5151.
- (15) (a) Badjic, J. D.; Balzani, V.; Credi, A.; Lowe, J. N.; Silvi, S.; Stoddart, J. F. *Chem.—Eur. J.* **2004**, *10*, 1926–1935. (b) Zhang, J.-N.; Li, H.; Zhou, W.; Yu, S.-L.; Qu, D.-H.; Tian, H. *Chem.—Eur. J.* **2013**, *19*, 17192–17200. (c) Maurer, C.; Baumgartner, B.; Pabisch, S.; Akbarzadeh, J.; Peterlik, H.; Schubert, U. *Dalton Trans.* **2014**, *43*, 950–957.
- (16) (a) Bayly, C. I.; Cieplak, P.; Cornell, W.; Kollman, P. A. *J. Phys. Chem.* **1993**, *97*, 10269–10280. (b) Wang, J.; Wolf, R. M.; Caldwell, J. W.; Kollman, P. A.; Case, D. A. *J. Comput. Chem.* **2004**, *25*, 1157–1174. (c) Lopes, J. N. C.; do Couto, P. C.; da Piedade, M. E. M. *J. Phys. Chem. A* **2006**, *110*, 13850–13856. (d) Hess, B.; Kutzner, C.; van der Spoel, D.; Lindahl, E. *J. Chem. Theory Comput.* **2008**, *4*, 435–447.
- (17) For the synthesis of compound **3**, see: Li, H.; Zhang, H.; Zhang, Q.; Zhang, Q.-W.; Qu, D.-H. *Org. Lett.* **2012**, *14*, 5900–5903.
- (18) (a) Cantrill, S. J.; Fulton, D. A.; Heiss, A. M.; Pease, A. R.; Stoddart, J. F.; White, A. J. P.; Williams, D. J. *Chem.—Eur. J.* **2000**, *6*, 2274–2287. (b) Arico, F.; Badjic, J. D.; Cantrill, S. J.; Flood, A. H.; Leung, K. C. F.; Liu, Y.; Stoddart, J. F. *Top. Curr. Chem.* **2005**, *249*, 203–259. (c) Yang, C.; Mori, T.; Origane, Y.; Ko, Y. H.; Selvapalam, N.; Kim, K.; Inoue, Y. *J. Am. Chem. Soc.* **2008**, *130*, 8574–8575.